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(54) Title: DIREACTIVE MESOGENIC COMPOUNDS AND INTERMEDIATES			
(57) Abstract <p>The invention relates to direactive mesogenic compounds or mixtures thereof comprising mesogene-containing molecule, said mesogens having two side chains attached thereto which contain a terminal polymerizable functional group, said mesogens and said functional groups being separated by at least two to twenty spacer atoms, wherein both spacer groups have different chain length.</p>			

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DIREACTIVE MESOGENIC COMPOUNDS AND INTERMEDIATES

The invention relates to

5 direactive mesogenic compounds or mixtures thereof obtainable by
a) treating a mesogenic diol of formula I.



10

in which

MG is a mesogenic group.

15 with a mixture of the halides of formula II and III



30

$$X^2-(CH_2)_n-R^b \quad (III)$$

in which

X₁ and X₂ are each independently Cl, Br or I

25 m and n are different integers between 1 and 20.

R^a and R^b are each independently groups selected from

-CH₂OH or -CH=CH₂

wherein

10

W is H, CH_3 or Cl

35 in the presence of a base, and

2

b) treating the resulting intermediate

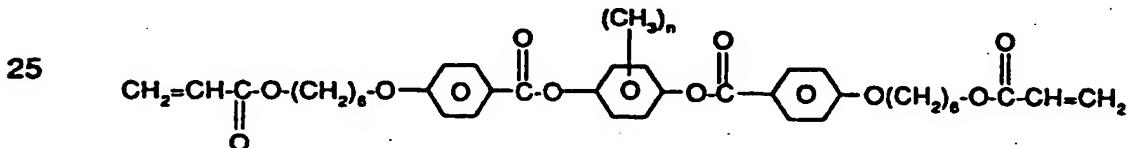
- in the case of R^a and R^b being -CH₂OH, with a vinyl derivative of formula CH₂=CW-(CO)_a-O- or a reactive derivative thereof, in which a is 0 or 1
- in the case of R^a and R^b being -CH=CWH with a perbenzoic acid.

10 The invention furthermore relates to the preparation of such compounds and to their use in electrooptical scattering systems and for the preparation of oriented liquid crystal polymers.

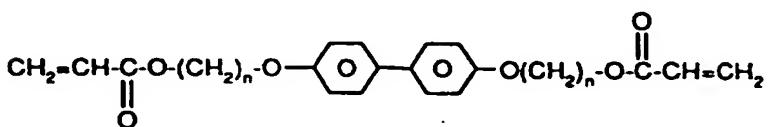
15 Reactive liquid crystal compounds can be polymerized in situ, whilst in their liquid crystal phase, to give highly crosslinked anisotropic films which can be used, for example, as polarizing beam splitters (see, for example, EP 0,428,213). Reactive liquid crystal compounds have furthermore been proposed for electrooptical scattering systems (see, for example, EP 0,451,905), cholesteric polarizers (e.g. EP 0,606,940) and

20 compensation films for STN displays (e.g. EP 0,423,881).

Reactive liquid crystal diesters of formula

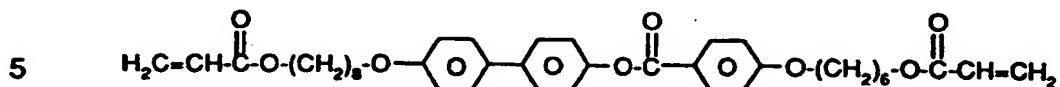


30 are mentioned in EP 0,261,712 (n=0), EP 0,331,233 (n=1). Reactive liquid crystal biphenyls of formula



35 are disclosed by EP 0,405,713.

The International Patent application WO 93/22397 discloses a compound of formula



These reactive liquid crystalline compounds often exhibit, however, rather high melting points disadvantageous values of the birefringence and comparable narrow mesophase ranges.

10

In view of the broad range of applications of reactive liquid crystal compounds it was desirable to have available further compounds of this type which fulfill the various requirements such as a reasonably low melting point, a high birefringence, a broad mesogenic range and 15 preferably an enantiotropic nematic range to a high degree.

20

It was an object of the present invention to provide new reactive liquid crystalline compounds with advantageous properties thus extending the pool of reactive liquid crystal compounds available to the expert. Other objects of the present invention can be taken from the following detailed specification.

25

The present invention thus relates to reactive mesogenic compounds or mixtures thereof obtainable by treating mesogenic diols of formula I, in particular those having a symmetric structure unit with a mixture of halides of formula II and III and to their use in electrooptical systems of scattering type and for the preparation of oriented liquid crystal polymers. The invention furthermore relates to the preparation of compounds according 30 to formula I.

30

Preferred embodiments of the present invention are:

a) Composition of direactive compounds comprising at least one compound of each formula IV, V and VI,

35



IV

4



V



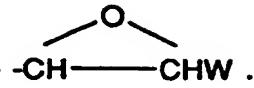
VI

5

in which MG, m and n have the meaning given, and

R¹ and R² are each independently

10



b) Diereactive compound or mixture thereof

15

in which

m - n ≥ 1, in particular 2, 3 or 4.

c) Diereactive compound or mixture thereof in which

20

MG is a mesogenic group of formula VII,



VII

25

in which

A¹ and A² are each independently

30

(a) 1,4-phenylene in which one or two CH groups may be replaced by N;

(b) 1,4-cyclohexylene in which one or two non-adjacent CH₂ groups may be replaced by -O- or one -CH- group may be replaced by -C(CN)-;

35

5

(c) naphthaline-2,6-diyl;

it being possible that group (a) is substituted by halogen
cyano or alkyl, alkoxy or alkanoyl with 1 to 6 C atoms,

5

Z¹ is each independently -COO-, -O-CO-, -CH₂-CH₂-,
-C≡C-, -CH₂O-, -OCH₂- or a single bond, and

o is 1, 2 or 3.

10

15

20

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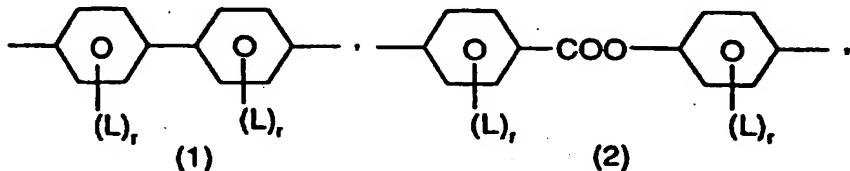
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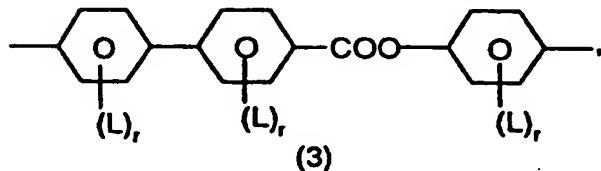
d) Directive compound or mixture thereof in which MG is selected from the structure elements (1) to (6).

5

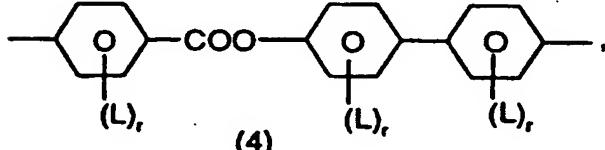


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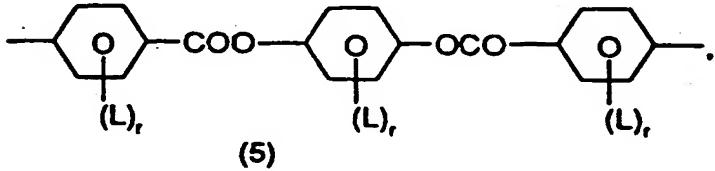
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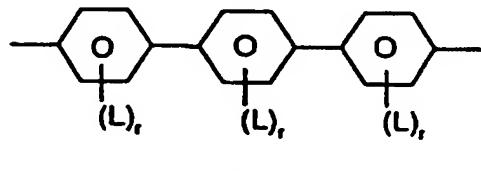
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20



25



in which

30

L is CH_3 , Cl, F, OCH_3 or $-\text{CO-CH}_3$, and

r is 0, 1, 2 or 4.

35

7

e) Directive compound or mixture thereof in which

n and **m** are given by the following table:

5	m	5	5	5	4	4
	n	2	3	4	2	3

A further aspect of the present invention are direactive compounds of formula IV A

10

$$\mathbf{R}^1\text{-}(\mathbf{CH}_2)_m\text{O}\text{-MG-O-}(\mathbf{CH}_2)_n\text{-R}^2 \quad (\text{IV A})$$

in which

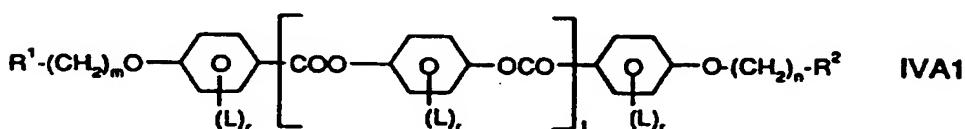
15 **R₁ and R₂** have the meaning given.

m and n are different integers between 2 and 10, and

20 MG is a mesogenic group, the core of which being symmetrical, preferably a structure element of formula (1), (5) or (6), in particular

directive compounds of the formula IVA1

25



30 in which

**R₁, R₂,
L and r have the meaning given.**

35

8

m and n are different integers between 2 and 10, and

t is 0 or 1.

5 Another aspect of the invention are the polymers prepared by polymerizing a monomer according to any of the preceding claims and

10 a chemical intermediate compounds or mixtures thereof useful in preparing direactive compounds or mixtures thereof according to any of the claim 1 to 8, comprising mesogene-containing molecules, said mesogenes having two side chains attached thereto that contain hydroxyl or vinyl group at the end thereof, said mesogenes and said functional groups being separated by at least two to twenty spacer atoms, wherein both spacer groups have different chain length.

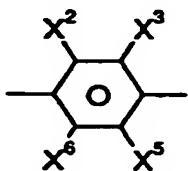
15

Above and below, the term reactive mesogenic compounds refers to reactive rod-like molecules which may be enantiotropic, monotropic or isotropic, preferably, however, enantiotropic or monotropic.

20

In the inventive compounds in which MG is a mesogenic group of formula VIII, A¹ and A² can be independently from each other an unsubstituted or a substituted 1,4-phenylene group of formula

25



30

X², X³, X⁵ and X⁶ can be independently from each other H, F, Cl, methyl or CN.

In the following, for the sake of simplicity, the following notation will be used:

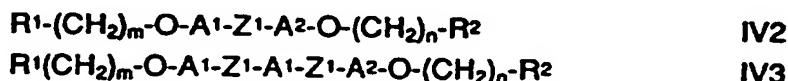
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9

Phe. 2 X² 3 X³ 5 X⁵ 6 X⁶ is a 1,4-phenylene group carrying in 2-position the group X², in 3-position the group X³ etc.; in case X², X³, X⁵ and/or X⁶, denote H, this will not be specified in above notation, i.e. only true substitutions will be listed. Thus Phe, for example, is an unsubstituted 1,4-phenylene group while Phe.2F 5 Cl is a 2-fluoro-5-chloro-1,4-phenylene group. Furthermore, Pyr is pyrimidine-2,5-diyl, Pyd is pyridine -2,5-diyl and Nap is a naphthalene-2,6-diyl group. The notation Pyr and Pyd in each case include the 2 possible positional isomers.

5

10 The compounds according to formula IV comprise 2- and 3-ring compounds (n=1 or 2) of formula IV2 and IV3:



15

In the 3-ring compounds of formula IV3, the ring groups A¹ can be chosen independently from each other.

20

Especially preferred is a smaller group of 2-ring compounds exhibiting the following structures for -A¹-Z¹-A²-:

25

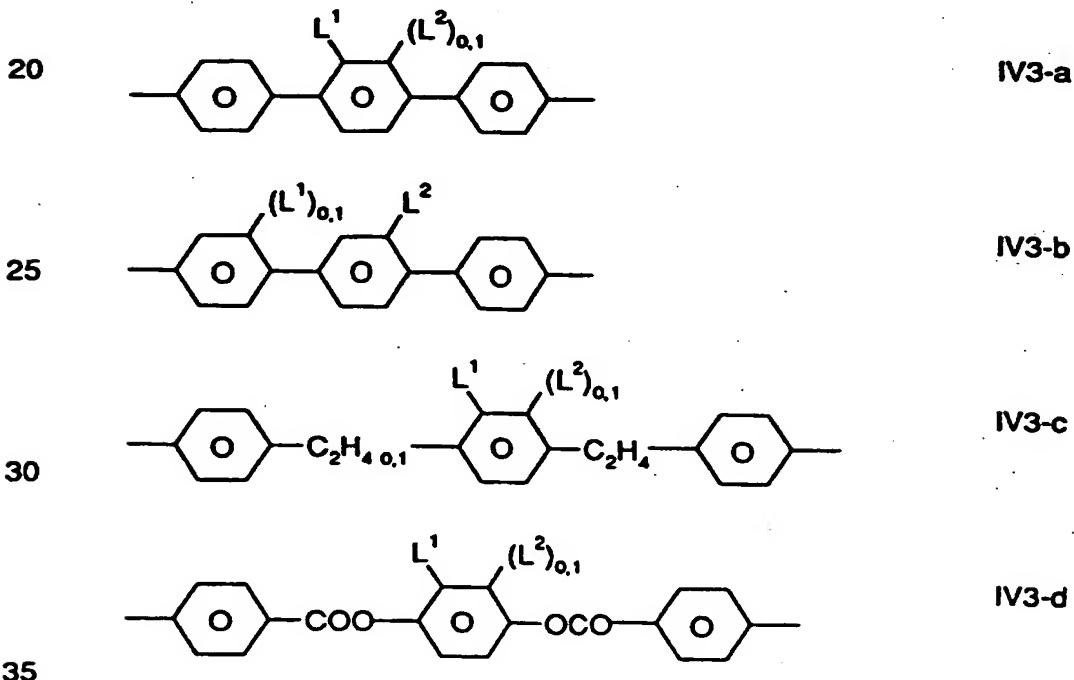
-Phe.2CH ₃ -Phe-	IV2-1
-Phe.3CH ₃ -Phe-	IV2-2
-Phe.2Cl-Phe-	IV2-3
-Phe.3Cl-Phe-	IV2-4
-Phe.2CN-Phe-	IV2-5
-Phe.3CN-Phe-	IV2-6
-Phe.2Cl3Cl-Phe-	IV2-7
-Phe.2Cl3F-Phe-	IV2-8
-Phe.2F-Phe-	IV2-9
-Phe.3F-Phe-	IV2-10
-Phe.-Phe-	IV2-11
-Phe.F-Nap-	IV2-12
-Phe.2Cl-Nap-	IV2-13

35

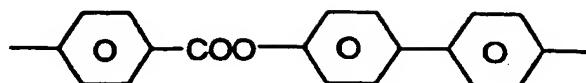
10

	-Phe.F-Nap-	IV2-14
	-Phe.3Cl-Nap-	IV2-15
	-Phe.2F-Pyr-	IV2-16
	-Phe.2F-Pyr-	IV2-17
5	-Phe.2CH ₃ -Pyd-	IV2-18
	-Phe.2Cl-Pyd-	IV2-19
	-Phe.F-CH ₂ CH ₂ -Phe-	IV2-20
	-Phe.3F-CH ₂ CH ₂ -Phe-	IV2-21
	-Phe.2Cl-CH ₂ CH ₂ -Phe-	IV2-22
10	-Phe.3Cl-CH ₂ CH ₂ -Phe-	IV2-23
	-Phe.2CN-CH ₂ CH ₂ -Phe-	IV2-24
	-Phe.3CN-CH ₂ CH ₂ -Phe-	IV2-25
	-Phe.2Cl3Cl-CH ₂ CH ₂ -Phe-	IV2-26
	-Phe.2Cl3F-CH ₂ CH ₂ -Phe-	IV2-27
15		

The 3-ring compounds according to formula IV3 preferably exhibit the following structures for -A₁-Z₁-A₁-Z₁-A₂:



II



IV3-e

5 In these structures, IV3-a to IV3-d, L¹ and L² denote independently from each other H, -Cl, -F, -CN and C₂H_{2+1-s} and, in particular, -Cl, -F, -CN, -CH₃, and/or -C₂H₅.

10 Especially preferred are the following patterns:

	-Phe-Phe.2CH ₃ -Phe-	IV3-a-1
	-Phe-Phe.2Cl-Phe-	IV3-a-2
	-Phe-Phe.2CN-Phe-	IV3-a-3
15	-Phe-Phe.2F-Phe-	IV3-a-4
	-Phe-Phe-Phe-	IV3-a-5
	-Phe-Phe.2C ₂ H ₅ -Phe-	IV3-a-6
	-Phe-Phe.3CH ₃ -Phe-	IV3-b-1
20	-Phe-Phe.3Cl-Phe-	IV3-b-2
	-Phe-Phe.3CN-Phe-	IV3-b-3
	-Phe-Phe.3F-Phe-	IV3-b-4
	-Phe-Phe.3C ₂ H ₅ -Phe-	IV3-b-5
25	-Phe.3F-Phe.3Cl-Phe-	IV3-b-6
	-Phe.3F-Phe.3CH ₃ -Phe-	IV3-b-7
	-Phe.3Cl-Phe.3Cl-Phe-	IV3-b-8
	-Phe.3Cl-Phe.3CH ₃ -Phe-	IV3-b-9
	-Phe-Phe.2Cl-Phe.3Cl-	IV3-b-10
30	-Phe-Phe.3Cl-Phe.3Cl-	IV3-b-11
	-Phe-Phe.2Cl-Phe.2Cl-	IV3-b-12
	-Phe-Phe.3Cl-Phe.2Cl-	IV3-b-13
	-Phe-Phe.2CH ₃ -Phe.3Cl-	IV3-b-14
	-Phe-Phe.3CH ₃ -Phe.3Cl-	IV3-b-15
35	-Phe-Phe.2CH ₃ -Phe.2Cl-	IV3-b-16

12

	-Phe-Phe.3CH ₃ -Phe.2Cl-	IV3-b-17
	-Phe-Phe.2F-Phe.3Cl-	IV3-b-18
	-Phe-Phe.3F-Phe.3Cl-	IV3-b-19
	-Phe-Phe.2F-Phe.2Cl-	IV3-b-20
5	-Phe-Phe.3F-Phe.2Cl-	IV3-b-21
	-Phe-Phe.2Cl-Phe.3CN-	IV3-b-22
	-Phe-Phe.3Cl-Phe.3CN-	IV3-b-23
	-Phe-Phe.2Cl-Phe.2CN-	IV3-b-24
	-Phe-Phe.3Cl-Phe.2CN-	IV3-b-25
10	-Phe-Phe.2CH ₃ -Phe.3CN-	IV3-b-26
	-Phe-Phe.3CH ₃ -Phe.3CN-	IV3-b-27
	-Phe-Phe.2CH ₃ -Phe.2CN-	IV3-b-28
	-Phe-Phe.3CH ₃ -Phe.2CN-	IV3-b-29
	-Phe-Phe.3F-Phe.3CN-	IV3-b-30
15	-Phe-Phe.2F-Phe.3CN-	IV3-b-31
	-Phe-Phe.3F-Phe.2CN-	IV3-b-32
	-Phe-Phe.2F-Phe.2CN-	IV3-b-33
	-Phe-Phe.2F-Phe.2F-	IV3-b-34
	-Phe-Phe.3F-Phe.3F-	IV3-b-35
20	-Phe-Phe.2CH ₃ -C ₂ H ₄ -Phe-	IV3-c-1
	-Phe-Phe.2Cl-C ₂ H ₄ -Phe-	IV3-c-2
	-Phe-Phe.2CN-C ₂ H ₄ -Phe-	IV3-c-3
	-Phe-Phe.2F-C ₂ H ₄ -Phe-	IV3-c-4
25	-Phe-Phe.2C ₂ H ₅ -C ₂ H ₄ -Phe-	IV3-c-5
	-Phe-Phe.2Cl3F-C ₂ H ₄ -Phe-	IV3-c-6
	-Phe-Phe.2Cl3Cl-C ₂ H ₄ -Phe-	IV3-c-7
	-Phe-C ₂ H ₄ -Phe.2CH ₃ -C ₂ H ₄ -Phe-	IV3-c-8
	-Phe-C ₂ H ₄ -Phe.2Cl-C ₂ H ₄ -Phe-	IV3-c-9
30	-Phe-C ₂ H ₄ -Phe.2CN-C ₂ H ₄ -Phe-	IV3-c-10
	-Phe-C ₂ H ₄ -Phe.F-C ₂ H ₄ -Phe-	IV3-c-11
	-Phe-C ₂ H ₄ -Phe.2OCF ₃ -C ₂ H ₄ -Phe-	IV3-c-12
	-Phe-COO-Phe.OCO-Phe-	IV3-d-1
35	-Phe-COO-Phe.2CH ₃ -OCO-Phe-	IV3-d-2

13

	-Phe-COO-Phe.3CH ₃ -OCO-Phe-	IV3-d-3
	-Phe-COO-Phe.2CH ₃ 3CH ₃ -OCO-Phe-	IV3-d-4
	-Phe-COO-Phe.2OCH ₃ -OCO-Phe-	IV3-d-5
	-Phe-COO-Phe.2Cl-OCO-Phe-	IV3-d-6
5	-Phe-COO-Phe.2F-OCO-Phe-	IV3-d-7
	-Phe-COO-Phe.2F3F-OCO-Phe-	IV3-d-8
	-Phe-COO-PhePhe-	IV3-e-1
10	-Phe-COO-Phe.2FPhe-	IV3-e-2
	-Phe-COO-Phe.3FPhe-	IV3-e-3
	-Phe-COO-PhePhe.2F-	IV3-e-4
	-Phe-COO-PhePhe.3F-	IV3-e-5
	-Phe.2F-COO-PhePhe-	IV3-e-6
	-Phe.3F-COO-PhePhe-	IV3-e-7

15

It was observed that the stability of 3-ring compounds wherein one of the 2 groups Z¹ is -COO- or -OCO- while the other denotes a single bond, can be increased if the compound is laterally di- or higher substituted, particularly di-substituted by -Cl, -F, -CN and/or -CH₃. Compounds of this type are preferred.

20

Especially preferred are further 3-ring compounds where both groups Z¹ are either -COO-, or -OCO- and at least one of the rings A¹, A^{1'} and A² are at least mono substituted.

25

In the compounds of formula IV R¹ is CH₂ = CW-COO-CH₂-, CH₂ =



CH-O-CH₂-, HW-C-C-, with W being H, Cl or alkyl with 1-5 C atoms and m being 1-7.

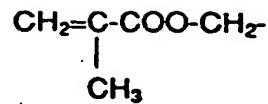
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Preferably, R¹ and R² is a vinyl group, an acrylate group, an epoxy group and especially preferred are the following means of R¹ and R²:

35

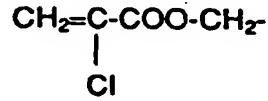
R¹⁻¹

14



R1-2

5

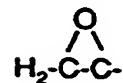


R1-3

10



R1-4



R1-5

with alkyl denoting C₁-C₃-alkyl and m being 1-5.

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15

The reaction methods mentioned are briefly summarized in the following synthetic tree:

Scheme I

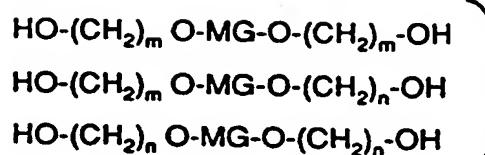
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10

+
Mixture of
Br-(CH₂)_m-OH/
Br-(CH₂)_n-OH
Butanone, K₂CO₃, heat

15



} Mixture

20

Butylvinyl ether
DCM
1,10-Phenanthroline
Pd (II)-Acetate



Mixture of
"diacrylates"

25

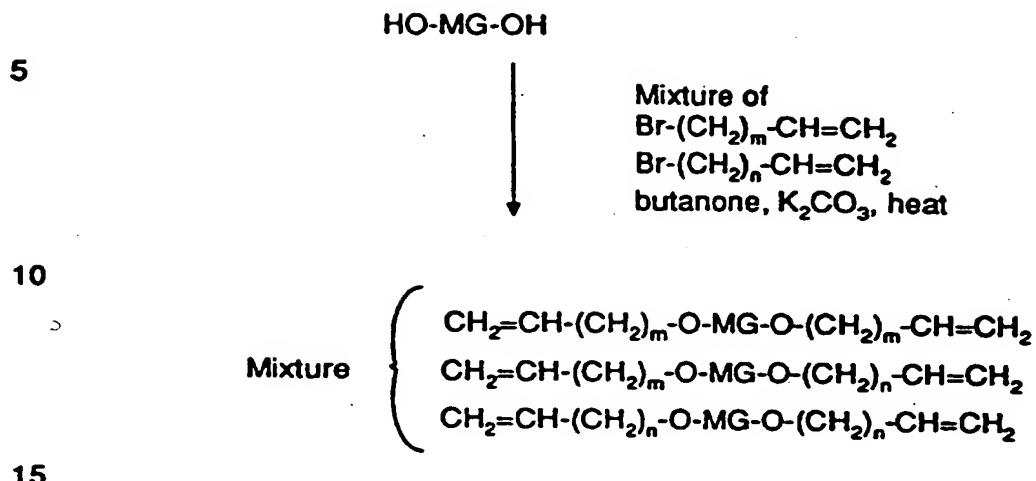
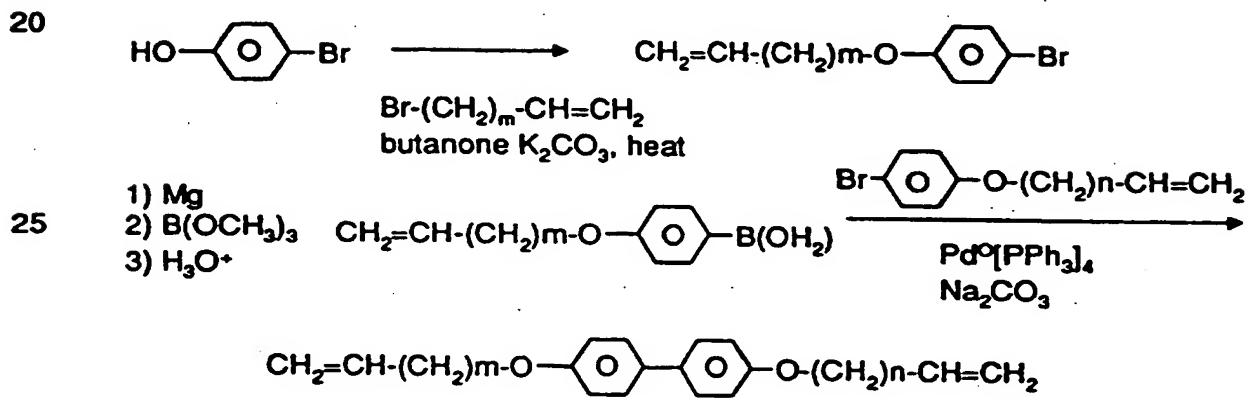
Mixture of
"Divinylethers"

DCM = dichloromethane

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35

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Scheme IIScheme III

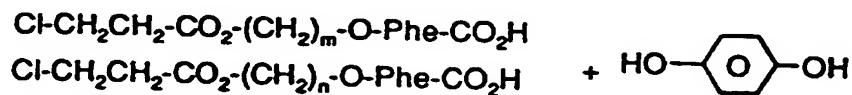
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Certain compounds or mixtures according to the invention are obtainable in a "one-pot-synthesis" as outlined in scheme IV:

5 Scheme IV

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1) TFAA/DCM
2) TEA/DCM

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TFAA = trifluoroacetic acid
TEA = triethylamine

25

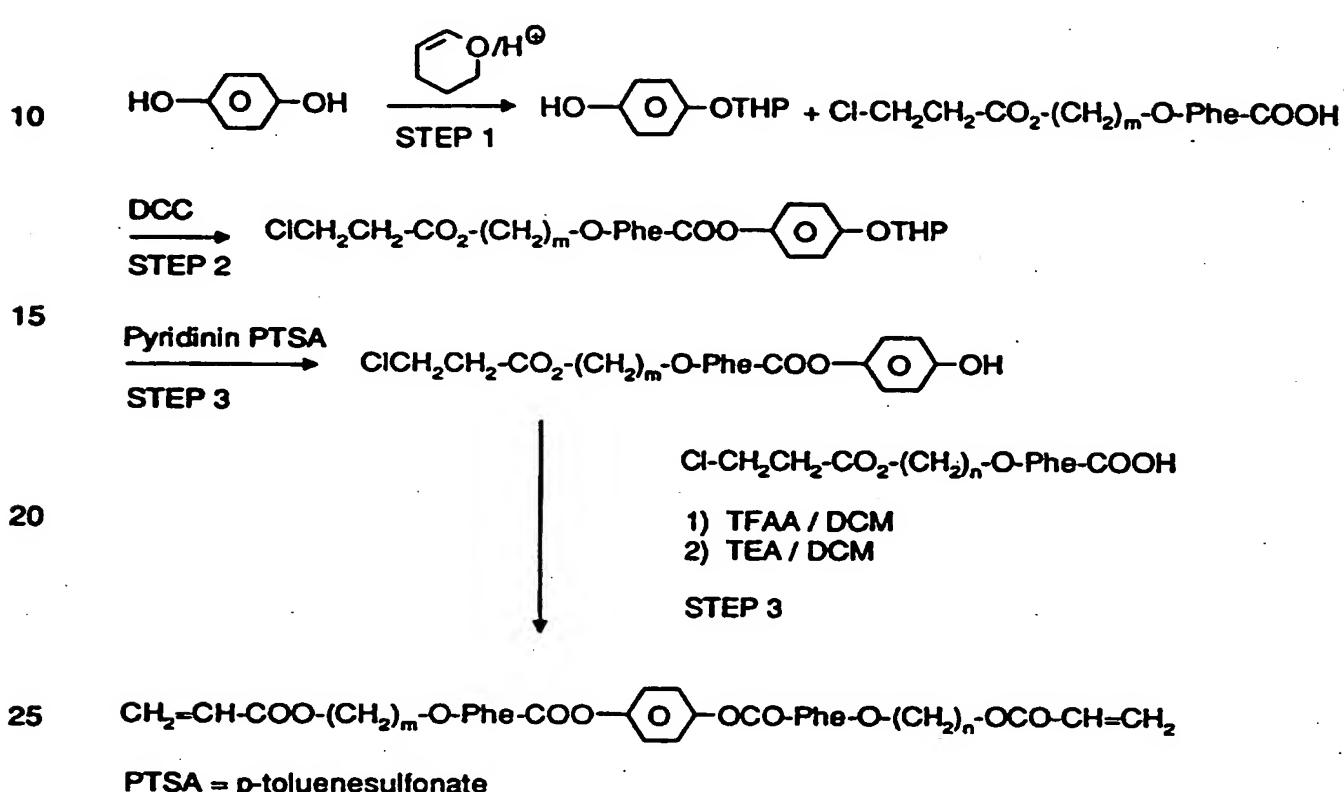
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Individual compounds of this type can be obtained according to schemes V to VII:

5 Scheme V

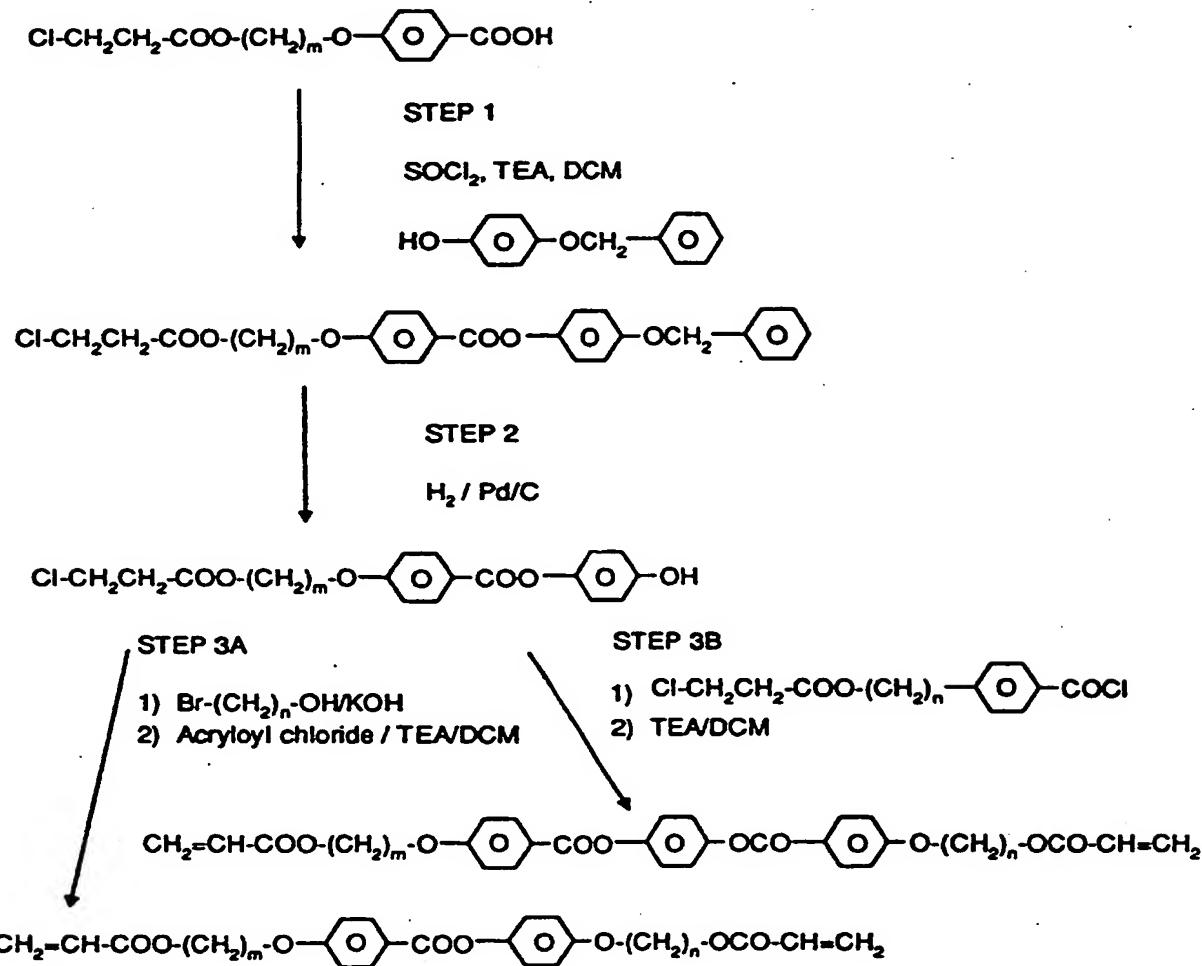


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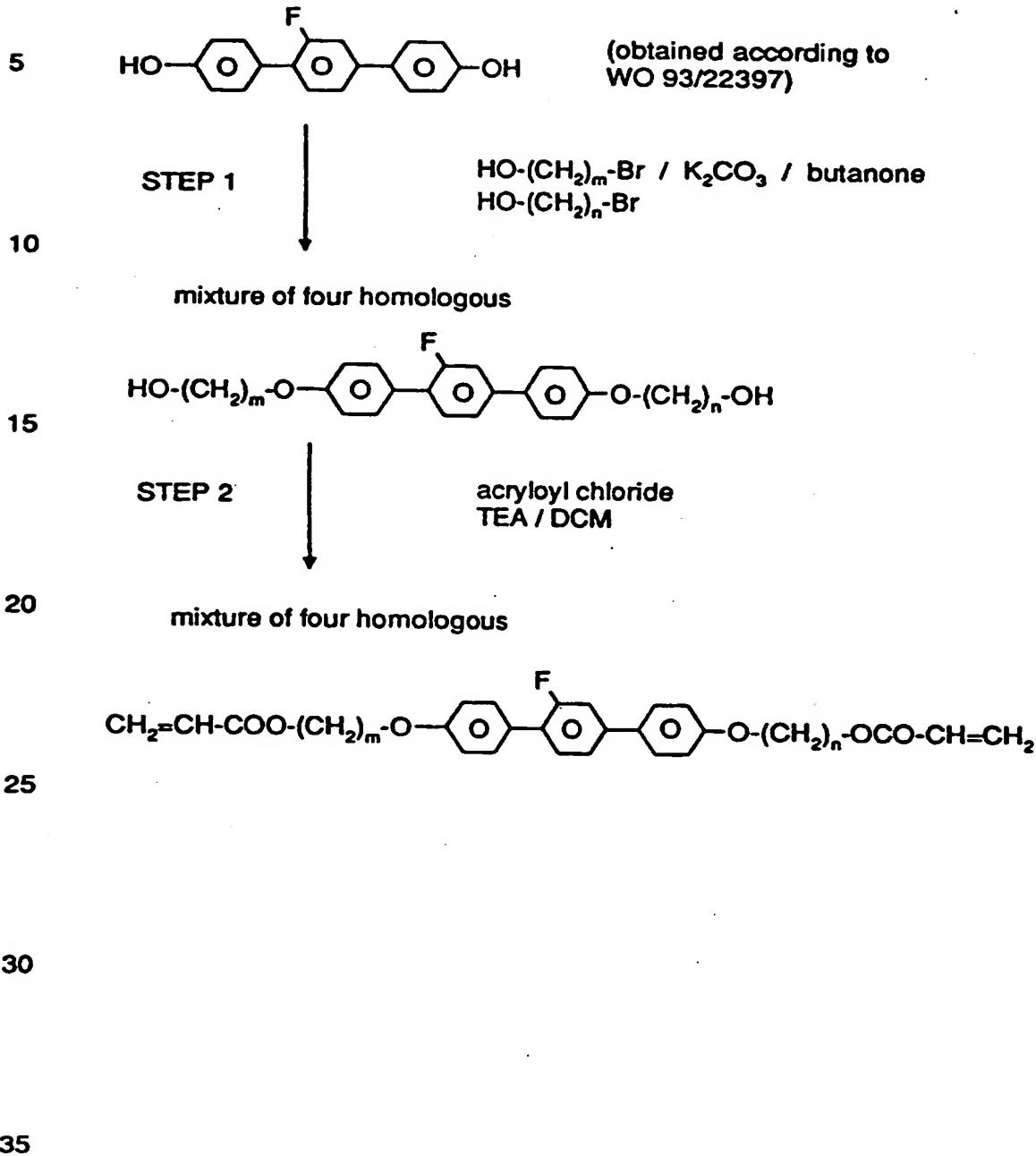
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Scheme VI



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Scheme VII



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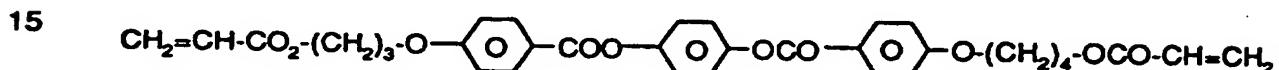
The reaction schemes mentioned above are to illustrate the invention without restricting it. The expert can choose other reaction methods without any inventive efforts.

5 In the following and in the preceding, all percentages given are percentages by weight. Temperatures are given in degrees Celsius.

10 The following examples are intended to illustrate the invention without restricting it.

Example 1

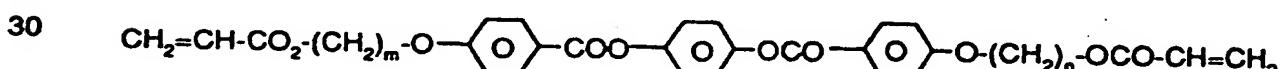
The reactive liquid crystalline compound (1)



20 is prepared via the sequence of reaction steps shown in scheme V. In step 4 of scheme V 2.2 moles TFAA are added dropwise to a solution 2 moles of the phenol obtained in step 3 of scheme V and 2 moles of the benzoic acid in 2 l of DCM.

25 The reaction mixture is stirred at room temperature for 16 hours. Then 5 moles of TEA in 1 l of DCM are added. The mixture is stirred for 16 hours. Aqueous work-up and column chromatography give (1) which shows K 111. S.

The following compounds are obtained analogously:



Compound No.	m	n	phase transition temperatures (°C)
(2)	3	5	K 76 N 160 I
(3)	4	5	K 67 N 165 I
5 (4)	4	6	K 78 S 82 N 159 I
(5)	5	6	K 78 S 80 N 162 I
Comp 1	4	4	K 105 N 164 I
Comp 2	5	5	K 91 N 167 I
Comp 3	6	6	K 105 (S 95) 152 I

10

Example 2

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A mixture of 1 mol Cl-CH2CH2-COO-(CH2)4-O-C(=O)c1ccccc1-COOH,

1 mol Cl-CH2CH2-COO-(CH2)4-O-C(=O)c1ccccc1-COOH, and 1 mol

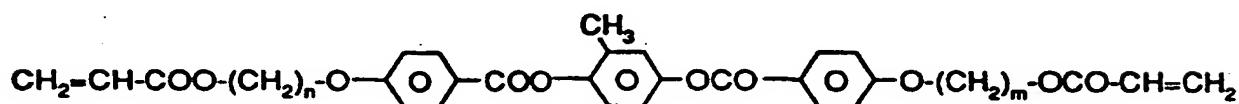
p-hydroquinone is treated with 22 moles of TFAA followed by 2.5 moles of TEA.

20

A mixture consisting of 1 part comp 1 and 1 part comp 3 and 2 parts of Compound No. (4) is obtained which shows a melting point of 56 °C and a clearing point of 163 °C.

Analogously a mixture of the following compounds is obtained:

25



30

n	m	parts
3	3	1
6	6	1
3	6	1
6	3	1

35

This composition shows a melting point below 30 °C and a clearing point of 119 °C.

Example 3

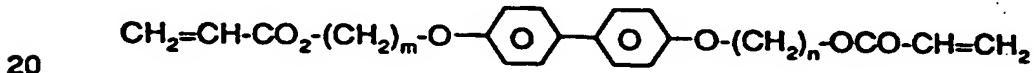
5 A mixture of 0.5 moles of biphenol, 1.3 moles of potassium hydroxide, 0.6 moles of 3-bromopropanol and 0.6 moles of 3 bromohexanol is heated in 2 litres of butanone for 16 hrs. After aqueous work-up a mixture of 1 part of $\text{HO}-(\text{CH}_2)_3-\text{O}-\text{C}_6\text{H}_4-\text{C}_6\text{H}_4-\text{O}-(\text{CH}_2)_3-\text{OH}$

10 1 part of $\text{HO}-(\text{CH}_2)_6-\text{O}-\text{C}_6\text{H}_4-\text{C}_6\text{H}_4-\text{O}-(\text{CH}_2)_6-\text{OH}$, and 2 parts of $\text{HO}-(\text{CH}_2)_6-\text{O}-\text{C}_6\text{H}_4-\text{C}_6\text{H}_4-\text{O}-(\text{CH}_2)_3-\text{OH}$

15 is obtained.

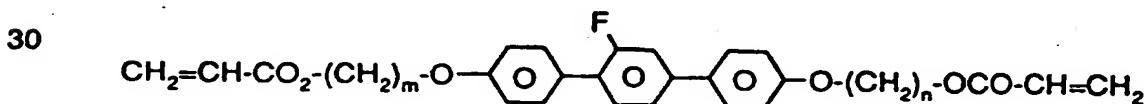
This mixture is treated with 1.2 moles of acryloyl chloride and 1.2 moles of TEA in 1.5 litres of dichloroethane DCM and refluxed for 3 hours.

15 After aqueous work-up the resulting reaction mixture is purified by column-chromatography to yield a mixture of diacrylates (6), (7) and (8).



Compound No.	m	n	parts
(6)	3	3	1
(7)	5	6	1
25 (8)	3	6	2

Analogously a mixture of compounds of formulae (9), (10), (11) and (12) is obtained



24

<u>Compound No.</u>	<u>m</u>	<u>n</u>	<u>parts</u>
	3	3	1
	6	6	1
5	3	6	1
	6	3	1

10

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20

25

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Claims

1. Diereactive mesogenic compounds or mixtures thereof obtainable
by

5

a) treating a mesogenic diol of formula I,

10



in which

MG is a mesogenic group,

15

with a mixture of the halids of formula II and III,



20

in which

X^1 and X^2 are each independently Cl, Br or J,

25

m and n are different integers between 1 and 20

R^a and R^b are each independently groups selected from



30

wherein

W is H, CH_3 or Cl.

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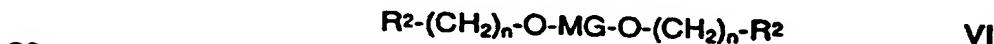
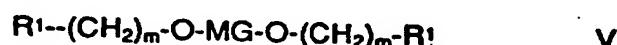
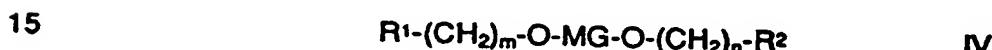
in the presence of a base, and

b) treating the resulting intermediate

5 - in the case of R^a and R^b being -CH₂OH, with a vinyl derivative of formula CH₂=CW-(CO)_a-O- or a reactive derivative thereof, in which a is 0 or 1

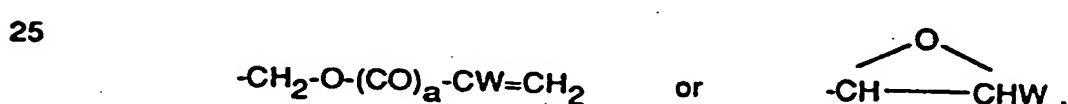
10 - in the case of R^a and R^b being -CH=CWH with a perbenzoic acid.

2. Composition of direactive compounds comprising at least one compound of each formula IV, V and VI,



in which MG, m and n have the meaning given, and

R¹ and R² are each independently



3. Direactive compound or mixture thereof according to claim 1 or 2

30 in which

m - n ≥ 1.

35

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4. Directive compound or mixture thereof according to any of the preceding claim in which

5 MG is a mesogenic group of formula VII,

-(A¹-Z¹)_o-A²-

VIII

in which

10 A¹ and A² are each independently

(a) 1,4-phenylene in which one or two CH groups may be replaced by N;

15 (b) 1,4-cyclohexylene in which one or two non-adjacent CH₂ groups may be replaced by -O- or one -CH- group may be replaced by -C(CN)-;

(c) naphthalene-2,6-diyl;

20 it being possible that group (a) is substituted by halogen cyano or alkyl with 1 to 6 C atoms,

25 Z¹ is each independently -COO-, -O-CO-, -CH₂-CH₂-, -C≡C-, -CH₂O-, -OCH₂- or a single bond, and

o is 1, 2 or 3.

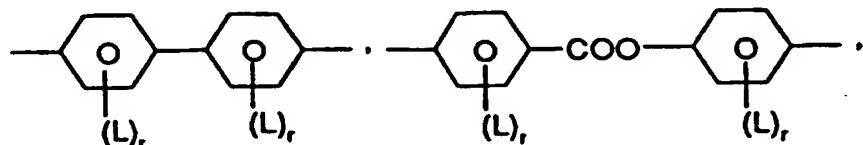
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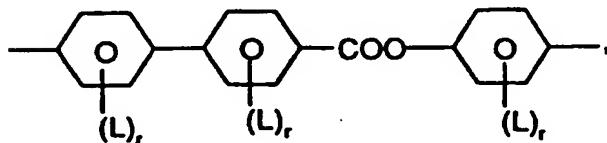
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5. Diereactive compound or mixture thereof according to any of the preceding claims in which MG is selected from

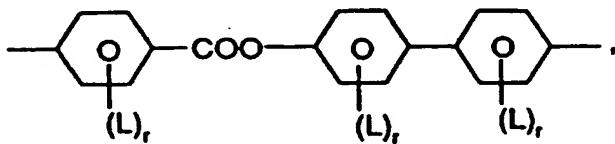
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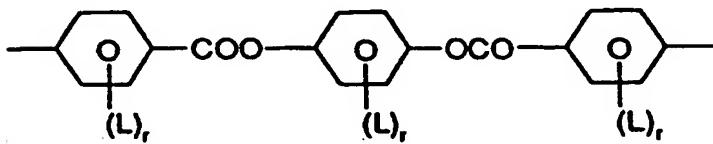
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in which

25 L is CH_3 , Cl, F, OCH_3 or $-\text{CO-CH}_3$, and

25 r is 0, 1, 2 or 4.

30 6. Diereactive compound or mixture thereof according to any of the preceding claims in which

35 n and m are given by the following table:

m	5	5	5	4	4
n	2	3	4	2	3

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7. Direactive compound of formula IV A

 $R^1-(CH_2)_mO-MG-O-(CH_2)_n-R^2$

(IVA)

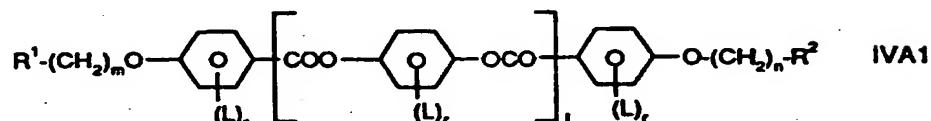
5 in which

R¹ and R² have the meaning given,

10 m and n are different integers between 2 and 10, and

MG is a mesogenic group, the core of which being
15 symmetrical.

8. Direactive compound according to claim 7 of the formula IVA1



20 in which

R¹, R²,
L and r have the meaning given,

25 m and n are different integers between 2 and 10, and

t is 0 or 1.

30 9. Polymers prepared by polymerizing a monomer according to any
of the preceding claims.35 10. Chemical intermediate compounds or mixtures thereof useful in
preparing direactive compounds or mixtures thereof according to
any of the claim 1 to 8, comprising mesogene-containing mole-
cules, said mesogenes having two side chains attached thereto

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that contain hydroxyl or vinyl group at the end thereof, said mesogenes and said functional groups being separated by at least two to twenty spacer atoms, wherein both spacer groups have different chain length.

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INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 96/00240

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6	C09K19/20	C09K19/12	C07C43/23	C07C69/92	C08F20/10
	C09K19/38				

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 C09K C07C C08F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO,A,93 22397 (MERCK PATENT GMBH) 11 November 1993 cited in the application see the whole document and particularly example 4.1 ---	1-10
X	GB,A,2 280 445 (MERCK PATENT GMBH) 1 February 1995 see claims 1-15 ---	1-9
Y	EP,A,0 261 712 (N.V. PHILIPS' GLOEILAMPENFABRIKEN) 30 March 1988 cited in the application see the whole document ---	1-9 -/-

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Date of the actual completion of the international search

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Date of mailing of the international search report

30 -05- 1996

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INTERNATIONAL SEARCH REPORT

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Information on patent family members

Final Application No

96/00240

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